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Note: Not all of the reactions will occur. For those that do not, write no reaction.

1. A piece of copper is dropped into a container of water. No reaction
2. Liquid bromine is added to a container of sodium iodide crystals. $\text{Br (l)} + 2\text{NaI(s)} \rightarrow 2\text{NaBr(s)} + \text{I}_2\text{(s)}$
3. An aluminum strip is immersed in a solution of silver nitrate.

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7. Free elements, no matter how complex the molecule, have an oxidation number (valence or charge) equal to zero. O, Fe, T, He < /1 1 ~ ~ L-td.
8. The following are diatomic or polyatomic elements in nature which must be committed to memory.

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Quiz on all Chemical Reactions - Max Study

The Ultimate Chemical Equations Handbook SnCO tin(II) carbonate sodium hydrogen carbonate NaHCO 1. 2. 3. 4. 5. 6. 8. 9. 10. 11. vanadium(V) oxide H₂O dihydrogen monoxide (NH) CO

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ammonium oxalate 4 224 polonium(VI) thiocyanate $\text{Po}(\text{SCN})_6$
tetraphosphorus decaoxide P_4O_{10} 12. 13. 15. 16. 17. 18. 19.
manganese(VII) oxide copper(II) dihydrogen phosphate

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KEY to 5 Reaction Prediction - Ipscience.com

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an alkali metal, its charge must be 1+ Oxygen is 2— but there are four of them, therefore, 4 times 2— equals 8— If 1+ and 8— are added The Ultimate Chemical Equations Handbook Answers Chapter 10

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