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Heating and changing pH levels are two ways to disrupt the shape of a protein. High T or pH levels that vary from the natural environment of the protein will break hydrogen bonds, ionic bonds, disulfide bridges, and hydrophobic interactions. Covalent bonds will remain undisturbed.

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14. Refer to Model 1. a. What happens to the shape of the hydrophobic tail in a phospholipid when a double bond is present in the carbon chain?

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It is a secondary structure where the protein coils because of hydrogen bonds between amino acids What is a beta pleated sheet 2 or more sequences of amino acids that are extended and aligned.

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6. Using your knowledge of protein structure, explain in detail the effect of exposing an enzyme to a pH outside of its optimal range. Include the effect on both enzyme structure and function. 7. At what pH values is lipase likely to be denatured? Justify your answer. 8. At what pH values is pepsin likely to be denatured? Justifr your answer. 9.

Mr. Schukow's Science Site - Homepage

⇒ The primary structure of a protein is the sequence of the amino acids that constitute it. Because of the nature of the peptide bond (cf. above), the backbone of a polypeptide will have a single primary amine at one end and a single carboxylic acid at the other end (at they do not take part in a peptide bond).

Introduction to proteins and protein structure [link] What ...

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