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3. Explain how the

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position of an element in the periodic table is related to the arrangement of electrons in its atoms.

(2) The elements group is the same as the number of electrons in its highest energy level

(1)

## **Atoms, Elements, Compounds and Mixtures part 1 Atoms ...**

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basic structure  
(protons, neutrons and  
electrons) of an atom  
To understand what  
valence electrons are  
and how you can

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Elements are arranged on the Periodic table in order of increasing atomic number where each element has one proton more than the element preceding it; Hydrogen has 1 proton, helium has 2 protons, lithium has 3 etc. The table is arranged in vertical columns called Groups numbered I - VIII and in rows called Periods; Elements in

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the same group have  
the same amount of  
electrons in their outer  
...

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alcohols, ethanol (boiling point  $78\text{ }^{\circ}\text{C}$ ) and butanol (boiling point  $118\text{ }^{\circ}\text{C}$ ). The apparatus is shown in Figure 1. Figure 1 1. 1 Complete the boxes in Figure 1 to identify the pieces of apparatus labelled A, B and C [3 marks]

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Tags: Question 2 .

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The sum of the number

of protons and

neutrons in an atom is

the mass number.

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### **Quiz - Quizizz**

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Atomic structure and the periodic table - Structured questions. Structured questions with 1, 2, 3 or 4 marks usually start with command words. If a question starts with the command word 'state ...

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Carbon; 2.4 Bulk &  
Answers  
Surface Properties; 3.  
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## **The Periodic Table | AQA GCSE Chemistry | Questions & Answers**

Atomic number =  
number of protons and  
electrons in an atom.

Atomic structure =  
protons/electrons.

Protons/electrons =  
atomic number. Atomic  
number(s) = found  
in/on the periodic

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is called the. A.

Electron. B. Neutron. C.

Proton . D. Nucleus. 2.

The ...

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